**Section 11.5: Electric Current**

**Tutorial 1 Practice, page 517**

1. **Given:** $Q = 0.20 \text{ mC}; \Delta t = 0.75 \text{ min}

**Required:** $I$

**Analysis:** $I = \frac{Q}{\Delta t}$

**Solution:** Convert time to seconds and charge to coulombs to get the answer in coulombs per second, or amperes:

$\Delta t = 0.75 \text{ min} \times \frac{60 \text{ s}}{1 \text{ min}}$

$\Delta t = 45 \text{ s}$

$Q = 0.20 \mu \text{C} \times \frac{1 \text{ C}}{1000 \mu \text{C}}$

$Q = 2.0 \times 10^{-4} \text{ C}$

$I = \frac{Q}{\Delta t}$

$I = \frac{2.0 \times 10^{-4} \text{ C}}{45 \text{ s}}$

$I = 4.4 \times 10^{-6} \text{ A}$

Convert the current to microamperes:

$I = 4.4 \times 10^{-6} \text{ A} \times \frac{1 \times 10^6 \mu \text{A}}{1 \text{ A}}$

$I = 4.4 \mu \text{A}$

**Statement:** The current travelling through the cellphone charger is $4.4 \times 10^{-6} \text{ A}$ or $4.4 \mu \text{A}$.

2. **Given:** $I = 15 \text{ A}; \Delta t = 24 \text{ h}$

**Required:** $Q$

**Analysis:** $I = \frac{Q}{\Delta t}$

**Solution:** Convert time to seconds to get the answer in ampere-seconds, or coulombs:

$\Delta t = 24 \text{ h} \times \frac{60 \text{ s}}{1 \text{ h}}$

$\Delta t = 86400 \text{ s}$ (one extra digit carried)

$I = \frac{Q}{\Delta t}$

$Q = I\Delta t$

$Q = (15 \text{ A})(86400 \text{ s})$

$Q = 1.3 \times 10^6 \text{ C}$

**Statement:** The number of electrons resulting from the current is $1.3 \times 10^6 \text{ C}$.

3. **Given:** $I = 250 \text{ mA}; Q = 1.7 \times 10^2 \text{ C}$

**Required:** $\Delta t$

**Analysis:** $I = \frac{Q}{\Delta t}$

**Solution:** Convert charge to coulombs, or ampere-seconds, and find time in seconds:

$Q = 1.7 \times 10^2 \text{ C}$

$I = \frac{Q}{\Delta t}$

$\Delta t = \frac{Q}{I}$

$\Delta t = \frac{1.7 \times 10^2 \text{ C}}{250 \text{ mA}}$

$\Delta t = 680 \text{ s}$

**Statement:** The time required to produce the charge is $680 \text{ s}$.

**Mini Investigation: How Much Current Can a Lemon Produce?, page 518**

A. Answers may vary. Sample answer:

When the lemon was connected, $0.5 \text{ V}$ was produced. When the LED load was added, the voltage dropped by $50 \text{ mV}$. The difference is negligible.

B. When more lemon cells are added in series, the brightness of the LED is increased.

**Section 11.5 Questions, page 518**

1. **Given:** $I = 4.4 \times 10^{-6} \text{ A}$

**Required:** $I$

**Solution:** Convert current to microamperes:

$I = 4.4 \times 10^{-6} \text{ A} \times \frac{1 \times 10^6 \mu \text{A}}{1 \text{ A}}$

$I = 4.4 \mu \text{A}$

**Statement:** The number of electrons resulting from the current is $1.3 \times 10^6 \text{ C}$.

2. **Given:** $Q = 0.20 \text{ mC}; \Delta t = 86400 \text{ s}$

**Required:** $I$

**Solution:** Convert time to seconds and charge to coulombs to get the answer in coulombs per second, or amperes:

$\Delta t = 86400 \text{ s}$

$Q = 0.20 \mu \text{C} \times \frac{1 \text{ C}}{1000 \mu \text{C}}$

$Q = 2.0 \times 10^{-4} \text{ C}$

$I = \frac{Q}{\Delta t}$

$I = \frac{2.0 \times 10^{-4} \text{ C}}{86400 \text{ s}}$

$I = 2.3 \times 10^{-8} \text{ A}$

**Statement:** The amount of charge travelling through the car battery is $2.3 \times 10^{-8} \text{ A}$.

3. **Given:** $I = 15 \text{ A}; \Delta t = 1.2 \text{ min}$

**Required:** $Q$

**Analysis:** $I = \frac{Q}{\Delta t}$

**Solution:** Convert time to seconds and find charge in coulombs:

$\Delta t = 1.2 \text{ min} \times \frac{60 \text{ s}}{1 \text{ min}}$

$\Delta t = 72 \text{ s}$

$I = \frac{Q}{\Delta t}$

$Q = I\Delta t$

$Q = (15 \text{ A})(72 \text{ s})$

$Q = 1.08 \times 10^3 \text{ C}$

**Statement:** The amount of charge travelling through the conducting wires is $1.08 \times 10^3 \text{ C}$.

4. **Given:** $I = 250 \text{ mA}; Q = 1.7 \times 10^2 \text{ C}$

**Required:** $\Delta t$

**Analysis:** $I = \frac{Q}{\Delta t}$

**Solution:** Convert current to microamperes and find time in seconds:

$I = 250 \text{ mA} \times \frac{1 \times 10^6 \mu \text{A}}{1 \text{ A}}$

$I = 2.5 \times 10^4 \mu \text{A}$

$Q = 1.7 \times 10^2 \text{ C}$

$\Delta t = \frac{Q}{I}$

$\Delta t = \frac{1.7 \times 10^2 \text{ C}}{2.5 \times 10^4 \mu \text{A}}$

$\Delta t = 680 \mu \text{A}$

**Statement:** The time required to produce the charge is $680 \mu \text{A}$. 

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**Solution:** Convert current to amperes to get the answer in coulombs per ampere, or seconds:

\[ I = 250 \text{ } \mu\text{A} \times \frac{1 \text{ A}}{1000 \text{ } \mu\text{A}} \]

\[ I = 0.25 \text{ A} \]

\[ I = \frac{Q}{\Delta t} \]

\[ \Delta t = \frac{Q}{I} \]

\[ = 1.7 \times 10^2 \text{ C} \]

\[ = \frac{1}{0.25} \text{ A} \]

\[ \Delta t = 680 \text{ s} \]

Convert the time to minutes:

\[ \Delta t = 680 \times \frac{1 \text{ min}}{60 \text{ s}} \]

\[ \Delta t = 11 \text{ min} \]

**Statement:** The battery can produce the current for 680 s or 11 min.

**5. Given:** \( Q = 150 \mu\text{C}; I = 0.21 \text{ mA} \)

**Required:** \( \Delta t \)

**Analysis:** \( I = \frac{Q}{\Delta t} \)

**Solution:** Convert charge to coulombs and the current to amperes to get the answer in coulombs per ampere, or seconds:

\[ Q = 150 \mu\text{C} \times \frac{1 \text{ C}}{1 \times 10^6 \mu\text{C}} \]

\[ Q = 1.5 \times 10^{-4} \text{ C} \]

\[ I = 0.21 \text{ mA} \times \frac{1 \text{ A}}{1000 \text{ } \mu\text{A}} \]

\[ I = 2.1 \times 10^{-4} \text{ A} \]

\[ I = \frac{Q}{\Delta t} \]

\[ \Delta t = \frac{Q}{I} \]

\[ = \frac{1.5 \times 10^{-4} \text{ C}}{2.1 \times 10^{-4} \text{ A}} \]

\[ \Delta t = 0.71 \text{ s} \]

**Statement:** The time required for the charge to pass through the LED light is 0.71 s.

**6.** First find the charge of the battery. Convert current to amperes and time to seconds to get the answer in coulombs.

\[ I = 2650 \mu\text{A} \times \frac{1 \text{ A}}{1000 \text{ } \mu\text{A}} \]

\[ I = 2.65 \text{ A} \]

\[ \Delta t = \frac{1 \text{ } \mu\text{C} \times 3600 \text{ s}}{1 \text{ } \mu\text{C}} \]

\[ \Delta t = 3600 \text{ s} \]

\[ I = \frac{Q}{\Delta t} \]

\[ Q = I \Delta t \]

\[ = (2.65 \text{ A})(3600 \text{ s}) \]

\[ Q = 9540 \text{ C} \] (two extra digits carried)

Now find the time it takes for 159 C of charge to deplete with a current of 883 mA. Convert current to amperes to get the answer in seconds.

\[ I = 833 \mu\text{A} \times \frac{1 \text{ A}}{1000 \text{ } \mu\text{A}} \]

\[ I = 0.833 \text{ A} \]

\[ I = \frac{Q}{\Delta t} \]

\[ \Delta t = \frac{Q}{I} \]

\[ = \frac{9540 \text{ C}}{0.833 \text{ A}} \]

\[ \Delta t = 1.15 \times 10^4 \text{ s} \] (two extra digits carried)

Convert the time to hours:

\[ \Delta t = 1.15 \times 10^4 \times \frac{1 \text{ h}}{3600 \text{ s}} \]

\[ \Delta t = 3 \text{ h} \]

So, the battery could produce a current of 883 mA for 3 h.

**7.** The student connected the ammeter in parallel so there is more than one path for the current to flow along. It is possible that the path passing through the ammeter has a much lower resistance than the path it is connected in parallel with, causing a large amount of current to take this path and resulting in a high reading.

**8.** Electric current is the conduction of free electrons in a material. If the material does not contain free electrons, then the material is not an electrical conductor. Therefore, no electric current can be produced in an non-conductor.

**9.** Electricians turn off the power to a circuit before working on it for safety. A current above 0.075 A is extremely dangerous when it flows into the body, and this is below a typical household circuit current rating. An electrician must always turn off the power to avoid accidental contact with an exposed wire or short-circuited electrical component.