

# 5PH4U-R WS#1

$$1.) \text{ error} = \frac{|9.72 - 9.81|}{9.81} \times 100\% \\ = 0.009174... \times 100\% \\ = 0.917\%$$

$$2) \text{a) diff} = \frac{|5.78 \times 10^{-34} - 7.29 \times 10^{-34}|}{\left( \frac{5.78 \times 10^{-34} + 7.29 \times 10^{-34}}{2} \right)} \\ = \frac{1.51 \times 10^{-34}}{6.535 \times 10^{-34}} \times 100\% \\ = 0.2310... \times 100\% \\ = 23.1\%$$

$$\text{b) error} = \frac{|5.78 \times 10^{-34} - 6.63 \times 10^{-34}|}{6.63 \times 10^{-34}} \times 100\% \\ = 0.1282... \times 100\% \\ = 12.8\%$$

$$\text{error} = \frac{|7.29 \times 10^{-34} - 6.63 \times 10^{-34}|}{6.63 \times 10^{-34}} \times 100\% \\ = 0.09954... \times 100\% \\ = 9.95\%$$

$$7.) \frac{1 \text{ g}}{x} = \frac{2.5 \times 10^{-9} \text{ m}}{1.0 \times 10^{-7} \text{ m}}$$

$$\frac{1}{x} = \frac{2.5 \times 10^{-9}}{1.0 \times 10^{-7}} \\ 2.5 \times 10^7 x = \frac{1.0 \times 10^{-7}}{2.5 \times 10^{-9}}$$

$$x = 40 \quad \boxed{x = 40 \text{ or } 4.0 \times 10^1 \text{ atoms}}$$

$$8.) \text{ Volume} = 10 \text{ cm} \times 1.2 \text{ cm} \times 15.6 \text{ cm} \\ = 187.2 \text{ cm}^3$$

$$1 \text{ cm}^3 = 2.7 \text{ g} \\ 187.2 \text{ cm}^3 = x$$

$$x = 505.44 \text{ g} \quad \boxed{x = 505.44 \text{ g or } 0.50544 \text{ kg}}$$

$$1.0 \text{ kg} = 2.2 \times 10^{25} \text{ atoms} \\ 0.50544 \text{ kg} = x$$

$$x = 1.1119... \times 10^{25} \\ \boxed{x = 1.1 \times 10^{25} \text{ atoms}}$$

$$9.) 1 \text{ star} = 2.0 \times 10^{30} \text{ kg} \\ 1.0 \times 10^{11} \text{ stars} = x$$

$$x = 2.0 \times 10^{41} \text{ kg}$$

$$10.) 1 \text{ s} = 2.50 \text{ C} \\ 600 \text{ s} = x$$

$$x = 1500 \text{ C}$$

$$1 \text{ C} = 6.24 \times 10^{18} \text{ e}^{\dagger} \\ 1500 \text{ C} = x$$

$$x = 9.36 \times 10^{21} \text{ e}^{\dagger} \quad \boxed{x = 9.36 \times 10^{21} \text{ e}^{\dagger}}$$